## Worksheet 15-3

Chemistry Worksheet Supplement - Molarity/molality
11. What is the molality of a solution composed of 3.25 g methanol $\left(\mathrm{CH}_{3} \mathrm{OH}\right)$ and 250 . g of water?
12. What is the molality of a solution made of 7.50 mg benzene $\left(\mathrm{C}_{6} \mathrm{H}_{6}\right)$ dissolved in 460.0 mg of toluene $\left(\mathrm{C}_{7} \mathrm{H}_{8}\right)$ ?
13. What is the molality of a solution of 5.0 g ethylene glycol $\left(\mathrm{C}_{2} \mathrm{H}_{6} \mathrm{O}_{2}\right)$ in 100 . g of water?
14. How many moles of NaCl are in 115 kilograms of a 0.250 m sodium chloride solution?
15. How many grams of $\mathrm{K}_{2} \mathrm{CO}_{3}$ are present in 3.0 kg of a 0.500 m (molal) solution of $\mathrm{K}_{2} \mathrm{CO}_{3}$ ?

